

Paper No. 11

Gujranwala Board
(First Group)ANNUAL
2016

ACCORDING TO THE NEW PAPER PATTERN OF ALL BOARDS

Roll No. (in Figures): (in Words):

Maximum Marks: 12

(OBJECTIVE TYPE)

Time Allowed : 15 Minutes

	A	B	C	D	Write correct option
1	(A)	(B)	(C)	(D)	
2	(A)	(B)	(C)	(D)	
3	(A)	(B)	(C)	(D)	
4	(A)	(B)	(C)	(D)	
5	(A)	(B)	(C)	(D)	
6	(A)	(B)	(C)	(D)	
7	(A)	(B)	(C)	(D)	
8	(A)	(B)	(C)	(D)	
9	(A)	(B)	(C)	(D)	
10	(A)	(B)	(C)	(D)	
11	(A)	(B)	(C)	(D)	
12	(A)	(B)	(C)	(D)	

NOTE: Four possible answers A, B, C and D to each question are given. The choice which you think is correct, fill that circle in front of that question with Marker or Pen ink. Cutting or filling two or more circles will result in zero mark in that question.

Q1.

12

- Metals can form ions carrying charges:
 - uni-positive
 - di-positive
 - tri-positive
 - all of these
- The oxidation number of chromium in $K_2Cr_2O_7$ is:
 - +2
 - +7
 - +14
 - +6
- How many times liquids are denser than gases?
 - 100 times
 - 1000 times
 - 10,000 times
 - 10 times
- Atoms react with each other because:
 - they attract each other
 - they are short of electrons
 - they want to attain stability
 - they want to disperse
- Mendeleev's Periodic Table was based on the:
 - electronic configuration
 - atomic mass
 - atomic number
 - completion of subshell
- Industrial Chemistry deals with manufacturing of compounds _____.
 - in the laboratory
 - on micro scale
 - on commercial scale
 - on economic scale
- The Molar Mass of H_2SO_4 is:
 - 98 gm
 - 98 amu
 - 9.8 gm
 - 9.8 amu
- When U-235 breaks up, it produces:
 - electrons
 - neutrons
 - protons
 - nothing
- Modern Periodic Table has ____ periods.
 - 3
 - 5
 - 7
 - 9
- A bond pair in covalent molecules usually has:
 - one electron
 - three electrons
 - four electrons
 - two electrons
- Which of the following is a "liquid in solid" solution?
 - sugar in water
 - butter
 - salt in water
 - smoke
- Oxidation number of all the elements in free state is:
 - 0
 - 1
 - 2
 - 3