D G KHAN FIRST GROUP

ANNUAL

12

ACCORDING TO THE NEW PAPER PATTERN OF ALL BOARDS

Roll No.(in Figures): -----(in Words): -----OBJECTIVE TYPE Maximum Marks: 12 Time Allowed: 15 Minutes 国国国国 0000 23 NOTE: Four possible answers A, B, C and D to each question are given. The choice which you think is correct, fill that circle in front of that question with Marker or Pen ink. Cutting or filling two or more circles will result in zero mark in that question. Q1. How many times liquids are denser than gases? I. (A) 100 time (B) 1000 time (C) 10,000 time (D) 100,000 time When a saturated solution is diluted it turns into: 2. (A) super saturated solution (B) colloidal solution (C) concentrated solution (D) unsaturated solution Which one of the following is a liquid in solid solution? (A) sugar in water (B) butter (C) salt in water (D) fog In the redox reaction between Zn and HCl the oxidizing agent is: (A) Zn (B) H+ (C) CF (D) H₂Spontaneous reaction takes place in. 5. (A) electrolytic cell (B) galvanic cell (C) nelson cell (D) down's cell Metals lose their electrons easily because: 6. (A) they are electronegative (B) they have electron affinity (C) they are alectropositive (D) good conductor of heat Which one the following element is found in most abundance in the earth curst? 7. (A) silicon (B) oxygen (C) hydrogen (D) aluminium When U-235 breaks up it produces: 8. (A) electron (B) neutron (C) protons (D) alpha particles 4th and 5th period of the long form of periodic table are called: 9. (A) short periods (B) normal periods (C) long periods (D) very long periods Point out the correct statement about electron affinity: (A) it is measured in KJ mol⁻¹ (B) it involve absorption of energy (C) it decreases in period (D) it increases in group 11. Which one of the following is the weakest force? (A) ionic force (B) covalent force (C) intermolecular force (D) co-ordinate covalent force

(C) co-ordinate covalent (D) metal

12. A bond formed between tow non-metals is:

(B) ionic

(A) covalent