## Faisalabad Board 2018 (Second Group)

Rol	No.(in Figures): (in Words): SUBJECTIVE TYPE Time Allower	
ivia.	(PART - I)	1 :1.45 Hours
Q2.		(5×2=10)
(i)	Define element and give at least two examples.	(0.2-10)
(ii)	Write the formula of sulphuric acid and sugar.	
(iii)		
(iv)	Write two properties of canal rays.	
(v)	Write the electronic configuration (in sub-shell) of sodium.	
	Define period and group in periodic table.	
7 -7	Which are long periods? How any elements are present in long periods?	
	)Why ionization energy decreases in a group from top to bottom?	
	Write short answers to any FIVE (5) questions:	(5×2=10)
(i)	What is single covalent bond? Explain with an example.	
(ii)	Differentiated between lone pair and bond pair of electrons	
(iii)	Define intermolecular forces and give an example also.	1
(iv)	Define intermolecular forces and give an example also.  Why the density of gases increase on cooling?	
(v)	Define allotropy and give an example also.	
(vi)	How will you prove that whether given solution is a colloidal solution or not?	
(vii)	Why benzene is insoluble in water?	
(viii	Define solvent and give an example also.	
Q4.	Write short answers to any FIVE (5) questions:	(5×2=10)
(i)	What is the difference between spontaneous and non-spontaneous reactions?	
(ii)	Define oxidizing agent with an example.	
(iii)	Write two examples of weak electrolytes.	
(iv)	Name the by-products produced in nelson's cell.	
(v)	Write two physical properties of metals.	1.
(vi)	Explain the trend of electropositivity in group of periodic table.	
(vii)	Write two uses of sodium.	
(viii	What is the significance of water for survival of life?	
	(PART - II)	
Note	Attempt any TWO questions.	(2×9=18)
Q5.	(a) What is Bohr's atomic theory? Give its postulates.	5
	(b) Write at least four differences between a compound and a mixture.	4
Q6.	(a) What is dipole-dipole interaction? Explain with example of HCI.	5
	(b) On what factors vapour pressure depends upon? Explain.	4
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How electrolytic refining of cooper in carried out? Explain.

Explain any four types of solution with examples.

(a)

(b)