

PAPER NO.
34

MULTAN
BOARD

SECOND GROUP

ANNUAL
2018

ACCORDING TO THE NEW PAPER PATTERN OF ALL BOARDS

Roll No.(in Figures):

(in Words):

Maximum Marks: 12

OBJECTIVE TYPE

Time Allowed : 15 Minutes

	A	B	C	D	Write correct option
1	(A)	(B)	(C)	(D)	
2	(A)	(B)	(C)	(D)	
3	(A)	(B)	(C)	(D)	
4	(A)	(B)	(C)	(D)	

	A	B	C	D	Write correct option
5	(A)	(B)	(C)	(D)	
6	(A)	(B)	(C)	(D)	
7	(A)	(B)	(C)	(D)	
8	(A)	(B)	(C)	(D)	

	A	B	C	D	Write correct option
9	(A)	(B)	(C)	(D)	
10	(A)	(B)	(C)	(D)	
11	(A)	(B)	(C)	(D)	
12	(A)	(B)	(C)	(D)	

NOTE: Four possible answers A, B, C and D to each question are given. The choice which you think is correct, fill that circle in front of that question with Marker or Pen ink. Cutting or filling two or more circles will result in zero mark in that question.

Q1.

12

- The oxidation number of chromium in $K_2Cr_2O_7$ is:
(A) +2 (B) +6 (C) +7 (D) +14
- The most common example of corrosion is:
(A) chemical decay (B) rusting of iron
(C) rusting of aluminium (D) rusting of tin
- Which of the following non-metals _____ is extremely hard.
(A) graphite (B) phosphorus (C) iodine (D) diamond
- The element is found in most abundance in earth's crust:
(A) oxygen (B) aluminium (C) silicon (D) argon
- _____ consists of four sub shells.
(A) K shell (B) L shell (C) M shell (D) N shell
- Mendeleev's periodic table was based upon:
(A) electronic configuration (B) atomic mass
(C) atomic number (D) completion of a subshell
- _____ elements are there in the first period of long form of periodic table.
(A) 2 (B) 3 (C) 4 (D) 6
- Transfer of electrons between atoms results in:
(A) metallic bonding (B) ionic bonding
(C) covalent bonding (D) coordinate covalent bonding
- _____ pair has polar covalent bond.
(A) O_2 and Cl_2 (B) H_2O and N_2 (C) H_2O and C_2H_2 (D) H_2O and HCl
- One atmospheric pressure is equal to _____ Pascal's.
(A) 101325 (B) 10325 (C) 106075 (D) 10523
- In soft drink the solvent is:
(A) benzene (B) water (C) milk (D) oil
- Molarity is the number of moles of solute dissolved in:
(A) 1 kg of solution (B) 100 g of solvent (C) 1 dm³ of solvent (D) 1 dm³ of solution