

FAISALABAD BOARD 2019

"Chemistry" Class 9th

Time : 15 Min

Objective Type (Group II)

Total Marks : 12

Note: You have four choices for each objective type question as A,B,C&D. The choice which you think is correct. Fill that circle in front of that question number. Use marker or pen to fill the circles. Cutting or filling two or more circles will result in zero marks in that question.

Q.1	Questions	(A)	(B)	(C)	(D)
(i)	Which is amorphous?	Diamond	Sodium chloride	Glass	Glucose
(ii)	Which is a liquid in solid solution?	Sugar in water	Butter	Salt in water	Fog
(iii)	Which is heterogeneous mixture?	Milk	Ink	Sugar solution	Paints
(iv)	The oxidation number of hydrogen in metal hydrides is:	+1	-1	+2	-2
(v)	Spontaneous chemical reaction takes place in:	Electrolytic cell	Galvanic cell	Nelson's cell	Downs cell
(vi)	Which is brittle?	Sodium	Aluminium	Magnesium	Selenium
(vii)	One amu is equivalent to:	$1.66 \times 10^{-24} \text{ mg}$	$1.66 \times 10^{-24} \text{ g}$	$1.66 \times 10^{-24} \text{ kg}$	$1.66 \times 10^{-23} \text{ g}$
(viii)	Which isotope is used for diagnosis of goiter in thyroid gland?	Sr-90	Co-60	I-131	C-14
(ix)	4th and 5th period of the long form of periodic table are called:	Short periods	Normal periods	Long periods	Very long periods
(x)	Which element has the highest electronegativity?	F	Cl	Br	I
(xi)	A bond formed between two non-metals is expected to be:	Covalent	Ionic	Coordinate covalent	Metallic
(xii)	The difference between electronegativity of hydrogen and chlorine is:		2	3	4