## FAISALABAD BOARD 2019

"Chemistry" Class 9th

Time: 15 Min

Objective Type (Group II)

Total Marks: 12

Note: You have four choices for each objective type question as A,B,C&D. The choice which you think is correct. Fit they circle in front of that question number. Use marker or pen to fill the circles.

two or more circles will result in zero marks in that question. Q.1 Questions (A) (B) (D) (C) (i) Which is amorphous? Diamond Sodium Glass Glucose chloride Which is a liquid in solid solution? (ii) Sugar in Butter Salt in water Fog water Which is heterogeneous mixture? (iii) Ink Paints Milk Sugar solution The oxidation number of hydrogen in (iv) +1 +2 -2 -1 metal hydrides is: Spontaneous chemical reaction takes Electrolytic Galvanic Nelson's cell Downs cell (v) place in: cell cell Which is brittle? Aluminium (vi) Sodium Magnesium Selenium (vii) One amu is equivalent ot: 1.66× 1.66× 1.66× 1.66× 10-24 mg 10-24 g  $10^{-24} kg$ 10-23 g Which isotope is used for diagnosis of Co-60 I-131 (viii) Sr-90 C-14 goiter in thyroid gland? 4th and 5th period of the long form of Short Normal Long Very long (ix) periodic table are called: periods periods periods periods Which element has the highest (x) Cl Br electronegativity? A bond formed between two Covalent Ionic (xi) Coordinal non-metals is expected to be: The difference between electronegativity (xii) of hydrogen and chlorine is: