			>C Chem	istry — 9
(	PAPER NO.	BO	ARD FIRST GRO APER PATTERN OF ALL BOARDS	ANNUAL 2018
Rol	l No.(in Figures):		(in Words):	
Ma	ximum Marks: 12	OBJECT	IVE TYPE	Time Allowed : 15 Minutes
	A B C D  A B	5 A B 6 A B 7 A B 8 A B		2 2 2 2
NO.		in front of that ques	tion with Marker or Pen i	e choice which you think is nk. Cutting or filling two or
Q1.				12
l.	One amu (Atomic Mass	Unit) is equivalent	to:	
	(A) 1.66 × 10 <sup>-24</sup> (mg)	(B) $1.66 \times 10^{-24}$ (g)	(C) $1.66 \times 10^{-24}$ (Kg)	(D) $1.66 \times 10^{-23}$ (g)
2.	The concept of orbit wa	s used by:	121 14	com
V	(A) J.J. Thomson	(B) Rutherford	(C) Bohr	(D) Plank
3.	Who presented law of o	ctaves?		Pa.
ď	(A) Dobereiner	(B) Mosely	(C) Newlands	(D) Mendeleev
4,	Mendeleev periodic tabl	e was base on:		3'
	(A) electronic configurati	on	(B) atomic mass	
	(C) atomic number		(D) completion of sub	shell
5.	Which pair of the molec	cules has double bo	nd?	
	(A) HCl and O <sub>2</sub>	B) N <sub>2</sub> and O <sub>2</sub>	(C) $C_2H_4$ and $O_2$	(D) $C_2H_2$ and $O_2$
6,	Which one of the follow	ing molecule is elec	trons deficient?	
	(A) NH <sub>3</sub>	(B) BF <sub>3</sub>	(C) N <sub>2</sub>	(D) O <sub>2</sub>
7.	How many times are the	e liquids denser tha	n gases?	Account of the control of
	(A) 100 times	B) 1000 times	(C) 10,000 times	(D) 100,000 times
8.	Mist is an example of so	lution:	4 2 2 20	A Section 1
	(A) liquid in gas	B) solid in gas	(C) gas in liquid	(D) gas in solid
9.	Tyndall effect is shown	by:		
	(A) sugar solution	B) paints	(C) jelly	(D) chalk solution
10.	The formula of rust is:			
		B) Fe <sub>2</sub> O <sub>3</sub>	(C) Fe(OH) <sub>3</sub> . nH <sub>2</sub> O	(D) Fe(OH) <sub>3</sub>
11.	Which is the most comn	on example of corr	3	2
	(A) chemical decay		(B) rusting of alumini	um
	(C) rusting of tin		(D) rusting of iron	
12.	Which one of the follow			
	(A) iodine	B) carbon	(C) phosphorus	(D) sulptuf

8.

9.