

PAPER NO.
44

BAHAWALPUR
BOARD

SECOND GROUP

ANNUAL
2018

ACCORDING TO THE NEW PAPER PATTERN OF ALL BOARDS

Roll No.(in Figures): (in Words):

Maximum Marks: 12

OBJECTIVE TYPE

Time Allowed : 15 Minutes

	A	B	C	D	Write correct option		A	B	C	D	Write correct option		A	B	C	D	Write correct option
1	(A)	(B)	(C)	(D)		5	(A)	(B)	(C)	(D)		9	(A)	(B)	(C)	(D)	
2	(A)	(B)	(C)	(D)		6	(A)	(B)	(C)	(D)		10	(A)	(B)	(C)	(D)	
3	(A)	(B)	(C)	(D)		7	(A)	(B)	(C)	(D)		11	(A)	(B)	(C)	(D)	
4	(A)	(B)	(C)	(D)		8	(A)	(B)	(C)	(D)		12	(A)	(B)	(C)	(D)	

NOTE: Four possible answers A, B, C and D to each question are given. The choice which you think is correct, fill that circle in front of that question with Marker or Pen ink. Cutting or filling two or more circles will result in zero mark in that question.

Q1.

12

- Which one of the following can be separated by physical means?
 - mixtures
 - elements
 - compounds
 - radicals
- Mendeleev periodic table was based on:
 - electronic configurations
 - atomic mass
 - atomic number
 - completion of subshell
- From left to right along the period which one of the following decreases:
 - atomic radius
 - ionization energy
 - electron affinity
 - electronegativity
- The concept of orbit was used by:
 - J.J. Thomson
 - Rutherford
 - Bohr
 - Planck
- Transfer of electron between atoms results in:
 - metallic bonding
 - ionic bonding
 - covalent bonding
 - coordinate covalent bonding
- Mist is an example of which solution:
 - liquid in gas
 - gas in liquid
 - solid in gas
 - gas in solid
- Which one of the following is crystalline solid?
 - rubber
 - plastic
 - glass
 - glucose
- Ice floats on water because:
 - ice is denser than water
 - water is denser than ice
 - ice molecules move randomly
 - water molecules move randomly
- Tyndall effect is shown by:
 - sugar solution
 - paints
 - jelly
 - chalk solution
- Non-metals are generally soft, but one of the following is extremely hard:
 - graphite
 - phosphorus
 - iodine
 - diamond
- Oxidation number of sulphur in H_2SO_4 :
 - +2
 - +7
 - +14
 - +6
- spontaneous reaction takes place in.
 - electrolytic cell
 - galvanic cell
 - Nelson cell
 - Daniell's cell